



Water desalination by membrane technology (RO) in southern Iran (Jask city)

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Abstract

Background: Reverse Osmosis (RO) is an increasingly common method of desalination. A full scale water desalination system by membrane technology (RO) evaluated in a southern city (Jask) in Iran.

Methods: First, data collection on water supply and network were performed. Analysis on most of the water quality parameters (Turbidity, pH, EC, Cl⁻, Na⁺, Alkalinity, Ca, Na, K, NO₃, NO₂, Fe, Mg, Mn, NH₄, PO₄, HCO₃⁻, SO₄²⁻ etc.) was performed as standard methods. The membranes of the RO in the desalination system were Poly-Amid (CSM type).

Results: The efficiency of the RO water desalination system was 94.16, 84.12, 92.00, and 96.17% respectively for Turbidity, Na⁺, Mg²⁺, SO₄²⁻. The result shows a significant difference between influent and effluent water of the RO system. The produced water is in agreement with national standard of drinking water. Furthermore, water exited from the RO system for TDS, Ca²⁺, and Mg²⁺ was less than minimum limit of the guideline.

Conclusion: The quality parameters of the water resource (EC, TDS, Na⁺ etc.) were higher than Iranian drinking water standards. The RO technology modified the quality of the water parameters

Keywords: Membrane technology, Reverse osmosis, Water desalination, Southern Iran

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Introduction

Dissolved salts removal from brackish and saline water for brackish and seawater treatment and also consumption was difficult and also expensive in the past, so saline water was not as a drinking water source. From 1950s, requirement of RO system to energy is less than the other desalination processes. It well ensures the global option for traditional usage. Isolation of dissolved salts marketing of RO system, RO membrane is a basic from a saline or brackish water, is called desalination. Ev treatment process for groundwater contaminated with ery water that contains Total Dissolved Salts (TDS) less than 1000 ppm, is called fresh water. Range of TDS Method of treatment for contaminated aquifer by RO for injection to desalination process is vary from 1000 process is the same as brackish water RO, to 60,000 ppm. Usually seawater contain TDS between 30,000-45,000 ppm, which can be removed by Reverse

Methods
This is an experimental and intermediate study on a full-scale water desalination system in Jask city of southern Iran (Figures 1, 2, 3). Use of membrane technology in desalination process, increased in recent years. (Recently, different industrial applications, use desalination process by RO membranes. This new technology, increased our potential for improvement of environmental protection and sustainable growth.)

Currently, RO method is considered as the best technology for brackish and seawater treatment and also consumption. It well ensures the global option for traditional usage. Isolation of dissolved salts marketing of RO system, RO membrane is a basic from a saline or brackish water, is called desalination. Ev treatment process for groundwater contaminated with ery water that contains Total Dissolved Salts (TDS) less than 1000 ppm, is called fresh water. Range of TDS Method of treatment for contaminated aquifer by RO for injection to desalination process is vary from 1000 process is the same as brackish water RO, to 60,000 ppm. Usually seawater contain TDS between 30,000-45,000 ppm, which can be removed by Reverse

EC, Cl, Na⁺, Alkalinity, Ca, Na, K, N₃NO₂, Fe, Mg, Mn, NH₄⁺, PO₄³⁻, HCO₃⁻, SO₄²⁻ etc.) was performed as standard methods. The membranes of the RO in the desalination system were Poly-Amid (CSM type). All experiments and preparation of the solutions were carried out based on the guidelines of a reference book titled “standard methods for water and wastewater experiments” (15).

Results

The [Tables 1 and 2](#) show, the water analysis for the inlet and outlet samples of the RO system. [Tables 1 and 2](#)



Figure 2. Full scale MBR system of Jask city - side a



Figure 1. Satellite image of Jask city



Figure 3. Full scale MBR system of Jask city - side b

Table 1. Water analysis for the inlet samples of the RO system

Parameters	Stage											
	1	2	3	4	5	6	7	8	9	10	11	12
Turbidity	0.53	2.240	1.70	1.64	19.100	6.57	1.78	8.32	0.507	0.54	0.47	1.77
pH	7.76	8.33	7.72	7.86	7.50	7.78	7.72	8.130	7.100	7.60	7.47	7.69
TDS	1441	1289	1379	1308	1484	1271.4	1540	1218	1603	1669	1326.5	1340.5
EC	2484	2223	2377	2255	2558	2192	2656	2100	2762.50	2878	2287	2311
T, °C	25	24	23	23	22.80	22.70	22.70	18.50	19	19	15	14.50
Total hardness	546	458	500	544	562	716	740	442	664	643	489	489
Flouride	0.78	0.42	0.48	0.61	0.610	0.570	0.62	0.31	0.92	0.80	0.59	0.46
Chloride	493	369	369	524.60	493	641	404	330	571.60	571.60	385	385
SO ₄	515	498	616	240	508	685	500	450	500	520	514	490
CO ₃	0	0	0	0	0	0	0	0	0	0	0	0
HCO ₃	117.4	113.20	110	127	120.60	119.60	121	126.80	116.20	116.80	130	131
NO ₂ ⁻	0.001	—	0	0.012	0	0	0.001	0.001	0.004	0.005	0	0.001
PO ₄	0.018	—	0.028	0.025	0.016	0.0068	0.012	0.0135	0.023	0.020	0.0098	0.0079
NO ₃	0.485	—	1.260	0.094	0.08	0.58	0.123	0.213	0.394	0.43	0.43	0.711
Ca ²⁺	151.5	111.60	124	137.70	153.100	185.200	190	95	164.30	158.30	115	117.50
Mg ²⁺	40.82	43.64	46.41	48.69	43.74	61.70	66.100	49.81	61.73	60.260	49.100	47.50
Na ⁺	320	275	285	285	340	385	300	250	320	330	300	300
K ⁺	10	10	9.810	10.180	10.100	11.50	10	12	10.43	10.120	10.200	10
Fe ²⁺	0.039	0.075	0.669	0.146	0.043	0.001	0	0.015	0.354	0.029	0.01	0
Mn ²⁺	0	0.003	0.005	0.015	0.130	0	0.111	0	0.001	0.005	0	0
NH ₃	0	0	0.308	0.183	0	0	0	0	0	0	0	0

Table 2. Water analysis for the outlet samples of the RO system

Parameters	Stage											
	1	2	3	4	5	6	7	8	9	10	11	12
Turbidity	0.278	0.157	0.139	0.493	0.492	0.04	0.140	0.143	0.23	0.145	0.201	0.18
PH	7.61	7.83	7.210	6.56	6.670	7.53	7.050	7.240	7.62	6.95	6.150	7.45
TDS	165	130	144.7	138.60	176	213	232	163	130	110.90	248	236
EC	300	235.70	263	252	319.1	386.60	422.50	297	236	201.6	450.30	429
T, °C	25	14.500	14.500	18.60	18.60	18.50	22.70	22.80	22	23	23	24
Total Hardness	33.200	15.400	39	25	24.20	64.40	46	41	17.40	28.40	45	64.40
Flouride	0.220	0.120	0.23	0.17	0.32	0.18	0.17	0.20	0.21	0.14	0.15	0.12
Chloride	80	59	60	75	58.100	63.130	46	49.54	59.100	42.74	107.8	49.53
SQ ₄	11	20.400	29.80	14.34	15.84	17.21	15.90	41.040	10.62	9.300	29.070	16.50
CO ₃	0	0	0	0	0	0	0	0	0	0	0	0
HCO ₃	26.80	16.60	19.60	20	17	25.20	19.80	24.40	18.80	23	20.20	16.50
NO ₂ ⁻	0	0.001	0.001	0.008	0.009	0.005	0.001	0.0008	0.013	0.0004	0	-
PO ₄	0.014	0.007	0.0063	0.016	0.013	0.007	0.006	0.005	0.013	0.024	0.029	-
NO ₃	0.331	0	0.42	0.286	0.394	0.2	0.4	0.3	0.394	0.22	4.918	-
Ca ²⁺	7.300	3.85	11.50	6.090	6.65	43.50	12.80	9.78	4.240	6.250	12.42	16.67
Mg ²⁺	3.64	1.40	2.50	2.38	1.80	10.72	10.40	4.030	1.65	3.010	3.40	5.46
Na ⁺	50	46	42	50	40	50	46	48	41	34	74	65
K ⁺	1.620	0.001	1.40	1.120	1.43	2.80	1.200	1.68	2.70	1.190	2.31	1.81
Fe ²⁺	0.231	0	0.02	0.041	0.028	0.03	0	0	0.006	0.147	0.147	0.111
Mn ²⁺	0	0	0	0.001	0.001	0	0	0	0	0.001	0.001	0
NH ₃	0	0	0	0	0	0	0	0	0	0.210	0.035	0

and Figures 46 indicate, this system can be getting the and turbidity. government standards of drinking water. RO system, in Some problems associated with using membranes may let and outlet parameters statistical results were indicated include short design life; membrane cleaning (backwash in Tables 3 and 4 Efficiency of the RO system that applied in this study, represented Figure 4 Range of operating pressures is different for brackish andouling. seawater, 250 to 400 and 800 to 1000 psi, respectively. As Figure 5 showed, except for nitrogen, phosphorus, The quality of feed water is a determining factor to decide iron and sulphate, removal efficiency for other param on the type of membrane process to use. Surface water (as compared to groundwater) represents the most variable turbidity was about 65%. The most efficiency removal water quality, particularly in terms of particle loadings related to Mg²⁺ and the minimum was related to PO₄

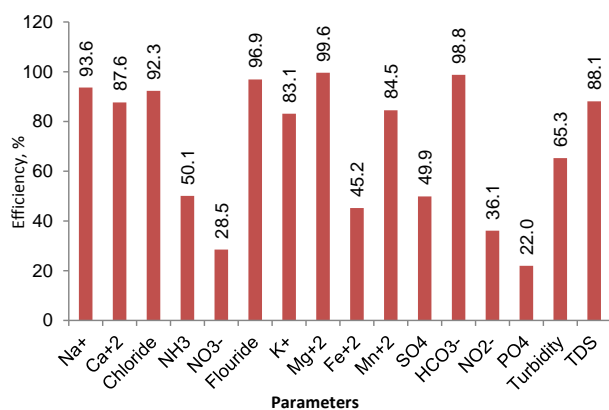


Figure 4. Removal efficiency of different parameters by RO system

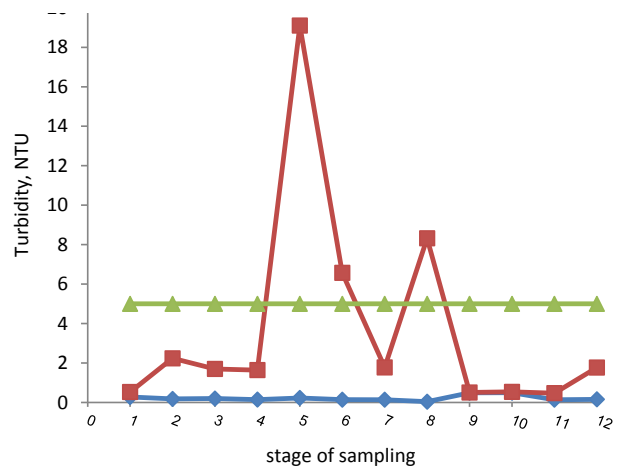


Figure 5. Turbidity status of the RO system in Jask city

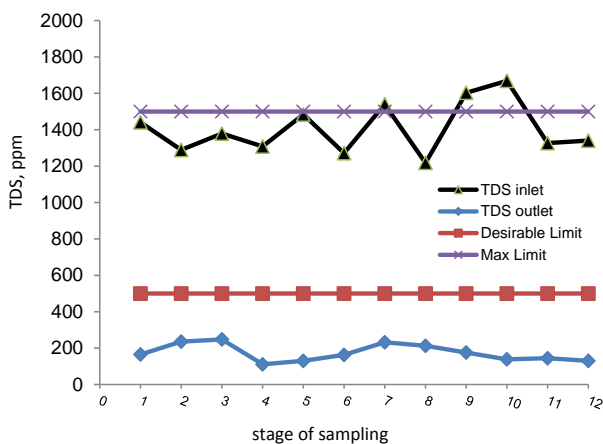


Figure 6. TDS status of the RO system in Jask city

Discussion

RO membrane systems have applied increasingly for sea water and brackish water. Although, permeate flux for brackish water is higher than seawater RO system, operational pressure and salt rejection are lower.

Saudi Arabia and the United States, with 26% and 17% have the first and second ranks in the global desalination capacity, respectively (16).

As the previous researches indicated, rejection of mono valent ions (such as CN^{\ominus}) and salt rejections by RO membranes, can be higher than 99% (16).

According to WHO and IR standards, the maximum allowable value for the parameters were illustrated in the

Tables 3 and 4. As shown in Figure 6 hardness in the feeds of Jask city plant was higher than the allowable limit recommended by WHO and IR standards. Treated water by RO system in the Jask city plant was agreeable with both WHO and IR standards. Moreover, efficiency of RO system for hardness reduction was found 98.5%. As presented in Tables 3 and 4 the influent and effluent of Jask city plant have lower turbidity, fluoride, magnesium and nitrate concentration levels than WHO and IR standards (17,18).

Applied RO desalination plant in Jask city is for brackish water. As Figure 4 indicates, removal efficiency of different parameters by RO system has an important role to deliver drinking water to the Jask city people.

Applied pressure is the basic agent for driving the RO system. The level of energy requirement for RO system severance is directly dependent on level of the salinity of the solution. Increasing of salinity can cause high-pressure requirement for drinking water production and high energy and cost (19,20). Jask city desalination plant is designed based on brackish water than seawater.

Conclusion

This research indicated that the efficiency of MBR system for removing the pollutants is effectively high and applicable in all seaside cities and other places which en-

counter with anions and cations so it is suggested that with respect to the deficiency of the water treatment system of this plant, MBR process can practically be used for the water treatment.

Application of this system in water systems with high coloured and turbidity is relatively low. Finally, we can conclude that this system is very suitable and ideal for water treatment with low concentration of color and turbidity such as groundwater.

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Ethical issues

We certify that all data collected during the study is presented in this manuscript and no data from the study has been or will be published separately.

Competing interests

The authors declare that they have no competing interests.

Authors' contributions

All authors participated in the design study, data acquisition, and interpretation. RM involved in the analyse and ZY wrote the manuscript.

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